J|A|C|S
ADDITIONS AND CORRECTIONS

Dramatic Increases in the Lifetime of the Er^{3+} Ion in a Molecular Complex Using a Perfluorinated Imidodiphosphonate Sensitizing Ligand [*J. Am. Chem. Soc.* **2005**, *127*, 524−525]. Gaetano Mancino, Andrew J. Ferguson, Andrew Beeby, Nicholas J. Long,* and Tim S. Jones*

We acknowledge that Drs. Zoe Pikramenou and Peter Glover (University of Birmingham) first prepared and provided a sample of the nonfluorinated $Er(tpip)_3$ compound for initial studies, though none of the work detailed in this paper has utilized that sample.

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10.1021/ja069982j
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Organocatalytic Direct Michael Reaction of Ketones and Aldehydes with *β*-Nitrostyrene in Brine [*J. Am. Chem. Soc.* **2006**, *128*, 4966−4967]. Nobuyuki Mase, Kaori Watanabe, Hidemi Yoda, Kunihiko Takabe,* Fujie Tanaka, and Carlos F. Barbas III

Page 4967. The following citation should be added as ref 5o:

(o) List, B.; Pojarliev, P.; Martin, H. J. *Org. Lett.* **2001**, *3*, 2423.

The authors apologize for this oversight.

JA069980Z

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Microporous Metal−Organic Frameworks Incorporating 1,4-Benzeneditetrazolate: Syntheses, Structures, and Hydrogen Storage Properties [*J. Am. Chem. Soc.* **2006**, *128*, 8904−8913]. Mircea Dincă, Anta F. Yu, and Jeffrey R. Long*

Page 8907. Footnote *a* in Table 1 should read as follows:

[a]Obtained with graphite-monochromated Mo Kα (λ = 0.71073 Å) radiation for **1**, **3**, **4**, and **5**, and synchrotron radiation (λ = 0.775 Å) for **2** and **6**.

Supporting Information Available: An incorrect spatial correction file was used for the X-ray detector at the Advanced Light Source of Lawrence Berkeley National Laboratory. This caused systematic errors in the lattice parameters and peak prediction and resulted in poor integration and loss of high-angle data for compounds **2** and **6**. A reintegration of the X-ray diffraction data for these compounds using the correct spatial correction file provided a more precise refinement of the crystal structures. Updated CIF files have therefore been submitted for compounds **2** and **6**. This material is available free of charge via the Internet at http://pubs.acs.org.

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Propargyl Glycosides as Stable Glycosyl Donors: Anomeric Activation and Glycoside Syntheses [*J. Am. Chem. Soc.* **2006**, *128*, 9620−9621]. Srinivas Hotha* and Sudhir Kashyap

We have now found that the transglycosylation reaction we described does not proceed in the presence of organic bases such as triethyl amine and diisopropylethyl amine. Furthermore, propargyl glycosides do not behave as glycosyl donors in dioxane·HCl or Et_2O·HCl. Interestingly, the transglycosylation reaction between the propargyl per-*O*-benzylated glucoside (**4a**) and menthol (**6a**) proceeded with 5 mol % of $HAuCl_4$ in acetonitrile at 60 °C, giving 45% of lactol **5a** and 30% of menthyl glucoside **5b**.

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An OFF−OFF−ON Fluorescence Sensor for Metal Ions in Stepwise Complex Formation of 2,3,5,6-Tetrakis(2-pyridyl)pyrazine with Metal Ions [*J. Am. Chem. Soc.* **2006**, *128*, 15976−15977]. Junpei Yuasa and Shunichi Fukuzumi*

Page 15977. Equation 2 is corrected to

$$(A - A_0) = (A_\infty - A_0) \times \frac{-K[\mathrm{M}^{n+}] + \sqrt{K^2[\mathrm{M}^{n+}]^2 + (4 - K)K[\mathrm{TPPZ}]_0(2[\mathrm{M}^{n+}] - [\mathrm{TPPZ}]_0)}}{(4 - K)[\mathrm{TPPZ}]_0} \quad (2)$$

(or I) (or I_∞)

The same correction is also necessary in Supporting Information S4, where the derivation of eq 2 is presented.

The plots of I/I_∞ and $\Delta Abs/\Delta Abs_\infty$ vs $[Sc^{3+}]$ for the titration of TPPZ by Sc^{3+} (Figure 3b,c) were fitted by the corrected form of eq 2. Thus, the formation constant of TPPZ−Sc^{3+} [$K = (1.4 \pm 0.1) \times 10^2$] is not changed.

Supporting Information Available: Derivation of eq 2 (S4, corrected). This material is available free of charge via the Internet at http://pubs.acs.org.

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